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the energy released in the breaking of the phosphate bonds come from when the water molecules are added to the ends of the adp and phosphate pieces, not from the actual "breaking" of phosphate bonds. adp + pi (+ energy)-> atp (+ water) because a phosphate group is added to the adp molecule. the reaction is endergonic. the energy released

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kinetic energy = ½ × mass × (speed)², or, in terms of algebraic symbols: If the mass is in kilograms and the speed is in meters per second, then kinetic energy is measured in Joules, the same as the unit of work. Critical Thinking Questions, part 2. Calculate the kinetic energy of a 1000 kg car traveling at 30 m/s, a typical speed on a highway.

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POGIL | Curricular Materials
A common topic in chemistry discussion groups and forums is about the use of the terms "spontaneous reaction" versus "thermodynamic favorability". This is a new activity for chemistry students who struggle with the correlation between changes in enthalpy, temperature, entropy, and the Gibbs free energy of a system; which relies on an analogy that most students will be familiar with.