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The number 10,210 has four significant figures because only the zero at the end is considered not significant. All of the digits in the number 10,005 are significant because the zeros are in between two nonzero numbers (Rule #3). Critical Thinking Questions 1. Verify that each of the following numbers contains four significant figures.

### **Information: Qualitative vs. Quantitative**

ChemQuest 1: Numbers in Chemistry.  
Information: Scientific Notation “  
Scientific notation ” is used to make very large or very small numbers easier to handle. For example the number 45,000,000,000,000,000 can be written as “ $4.5 \times 10^{16}$  ”.

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Period 2 By: khoi and Duy

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In chemistry and spectroscopy,  $l = 0$  is called an s orbital,  $l = 1$  a p orbital,  $l = 2$  a d orbital, and  $l = 3$  an f orbital. The value of  $l$  ranges from 0 to  $n - 1$  because the first p orbital ( $l = 1$ ) appears in the second electron shell ( $n =$

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2), the first d orbital ( $\ell = 2$ ) appears in the third shell ( $n = 3$ ), and so on.

## **Quantum Numbers | Introduction to Chemistry**

Consider the Arrhenius parameters for two first-order reaction: Reaction (1)  $E_a = 35 \text{ kJ/mol}$ ,  $A = 1 \times 10^{10} \text{ s}^{-1}$  View Answer You pour some water from a pitcher into a bucket located 0.55 m ...

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are in one km? . we saw in the last  
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For instructors, they are a good resource for homework, quiz, or test questions, or practice for the AP Chemistry test. Quiz

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