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YHS Chemistry Second Semester Final Exam Study Guide 1. What mass of hydrogen gas can be produced by reacting 6.92 g of sodium in excess water? The reaction is: 2. Calculate the masses of silver chloride and dihydrogen monosulfide formed when 85.6 g of silver sulfide reacts with excess hydrochloric acid, HCl. (double replacement reaction) a. AgCl = 99.0g b. H₂S = 11.8g 3.

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These guides are closely aligned with chapters 12-17 and 21 and were designed for use in the second semester sequence of general chemistry. They can be used in a flipped-style classroom where students complete them before the lecture. Or they can be used to reinforce important topics learned in class.

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