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electromagnetic
radiation. form of
energy that exhibits
wavelike behavior as it
travels through space
such as light.

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Choice 50% Name 10) 12)
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21) 22) 23) Problems
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your work for any
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3 out of 13 pages.
Chemistry (Chapter 5
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Answers
Match each item with the correct statement below. a. atomic orbital
d. ground state b. aufbau principle e. Pauli exclusion principle c. electron configuration f.

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lorelei_edwards. Words

to know. Terms in this

set (19) Orbital. A

region of space around

the nucleus where an

electron is likely to be

found. Orbit. TO MOVE

AROUND ANOTHER

OBJECT ALONG A PATH.

Chemistry Chapter 5

Test - Electrons in

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___ 5. Mendeleev is

credited with

developing the first

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Answers
successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ____ 6.

Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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TEST Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question. ... The most useful source of general information about the elements for anyone associated with chemistry is a a. calculator. c. ... Essay 65. Describe the differences between Mendeleev's periodic table and the ...

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1) A sample of argon gas is sealed in a container. The volume

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Answers
of the container is doubled at a constant temperature. What happens to the pressure inside the container? A) It cannot be predicted. B) It is squared. C) It is doubled. D) It is halved. E) It does not change.

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(Chapter 5) Class
Set Multiple Choice
(50%)**

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Electrons in Atoms

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