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Science Notes Chapter 2 Acids Bases and Salts. According to new CBSE Exam Pattern, MCQ Questions for Class 10 Science pdf Carries 20 Marks. CBSE Class 10 Science Notes Chapter 2 Acids Bases and Salts. Indicators: Indicators are substances which indicate the acidic or basic nature of the solution by the colour change.

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Acids Bases And

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NCERT Solutions for Class 10 Chapter 2 Science Acids bases and salts. In this chapter, students get hold of basic knowledge on Acids bases and salts. In this chapter, various chemical properties of acids, bases and salts and the reaction of them with metals, non-metals are studied.

NCERT Solutions for class 10 chapter 2

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Acids Bases And **Salts**

Acids, bases and salt

(ii) Acids react with metal carbonate and metals hydrogen carbonate to give salt, water and carbon dioxide. (iii) Acids react with bases to give salt and water. This reaction is called neutralization reaction. (iv) Acids react with metals oxides to give salt and water.

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Science experts in an easy to remember format, covering all syllabus of CBSE, KVPY, NTSE, Olympiads, NCERT & other Competitive Exams.

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Chapter 2 Acids, bases,
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Question 1 You have
been provided with
three test tubes.

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Chapter - 2, Acids
Bases and Salts Notes.

- These are the substances which have sour in taste.
- They turns blue litmus solution to red.
- These substances are bitter in taste.
- They turns red litmus solution to blue.
- They give OH ions in aqueous solution.

There are the

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Salts
substance that changes their colour / smell in different types of substances.

Chapter - 2, Acids Bases and Salts

Notes

Reaction of Acids and Bases with each other

- Acids and Bases react to form salt and water. Acid + Base → Salt + H₂O •

Neutralisation

Reaction: Reaction of acid with a base is

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called as neutralization
reaction.

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MCQs from CBSE Class
10 Science Chapter 2:
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1. Some fruits like
mango, lemon, raw

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grapes, orange, etc., have a sour taste due to the presence of: a. Acetic acid b. Citric acid

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Online Test of Chapter 2 Acid, Bases and Salt MCQ 1 Science| Class 10th Questions: 1. Basic characteristics of Acid is/are? i) Sour taste ii) Turns blue

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litmus red iii)Turns red

litmus blue iv)Both (i)
and (ii) 2.Basic

Characteristics of Base
is/are? i)Sour taste

ii)Turns red litmus blue

iii) urns blue litmus red

3.Which acid do we...

**Ch 2 Acid, Bases and
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Class 10th ...**

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**Chapter 2 Acids,
Bases and Salts -
Lakhmir Singh and**

...

For example,
Hydrochloric acid (HCl),
Sulphuric acid (H_2SO_4), Nitric acid (HNO_3).
Bases - Bases are
compounds which give

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Salts
hydroxide ion in water solution. For example, Sodium hydroxide (NaOH), Potassium hydroxide (KOH), Calcium hydroxide (Ca(OH)₂). Salts - Salts are compounds obtained by replacement of H from acid by a metal.

Chapter Notes: Acids Bases and Salts - Class 10 Science ...

Video Timeline:

0:00:00 start 0:00:21

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Difference between
Acids & bases 0:10:38

what are indicators?

0:17:07 Universal

Indicators 0:21:11 Salts

0:24:51 a) Sod...

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